

# B CONSTANTS & CONVERSION FACTORS

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## B.1. UNIVERSAL CONSTANTS

[from US National Institute of Standards and Technology (NIST), based on 2014 CODATA]

$c_0 = 2.99792458 \times 10^8 \text{ m}\cdot\text{s}^{-1}$  = speed of light in a vacuum

$c_1 = 3.741771790 \times 10^8 \text{ W}\cdot\text{m}^{-2}\cdot\mu\text{m}^4$  = first radiation constant (in Planck's law)

$c_{1B} = 1.191042953 \times 10^8 \text{ W}\cdot\text{m}^{-2}\cdot\mu\text{m}^4\cdot\text{sr}^{-1}$  = first radiation constant for spectral radiance

$c_2 = 1.43877736 \times 10^4 \mu\text{m}\cdot\text{K}$  = second radiation constant (in Planck's law)

$G = 6.67408 \times 10^{-11} \text{ m}^3\cdot\text{s}^{-2}\cdot\text{kg}^{-1}$  = Newtonian gravitational constant

$h = 6.626070040 \times 10^{-34} \text{ J}\cdot\text{s}$  = Planck constant

$k_B = 1.38064852 \times 10^{-23} \text{ J}\cdot\text{K}^{-1}\cdot\text{molecule}^{-1}$  = Boltzmann constant

$N_A = 6.022140857 \times 10^{23} \text{ mol}^{-1}$  = Avogadro constant

$\sigma_{SB} = 5.670367 \times 10^{-8} \text{ W}\cdot\text{m}^{-2}\cdot\text{K}^{-4}$  = Stefan-Boltzmann constant

$T = -273.15^\circ\text{C} = 0 \text{ K}$  = absolute zero (not considered a true universal constant)

## B.2. MATH CONSTANTS

[from *CRC Handbook of Chemistry and Physics*]

$e = 2.718281828459$  = base of natural logarithms

$1/e = 0.367879441$  = e-folding ratio

$\pi = 3.141592653589793238462643$  = pi

$\text{sqrt}(2) = 1.414213562373095$



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## B.3. EARTH CHARACTERISTICS

$1^\circ$  latitude = 111 km = 60 nautical miles (nm) [Caution: This relationship does NOT hold for degrees longitude.]

$a = 149.598 \text{ Gm}$  = semi-major axis of Earth orbit

$A = 0.306$  = Bond albedo (NASA 2015)

$A = 0.367$  = visual geometric albedo (NASA 2015)

$b = 149.090 \text{ Gm}$  = semi-minor axis of Earth orbit

$d = 149.5978707 \text{ Gm}$  = average sun-Earth distance

= 1 Astronomical Unit (AU) (NASA 2015)

$d_{\text{aphelion}} = 152.10 \text{ Gm}$  = furthest sun-Earth distance, which occurs about 4 July (NASA 2015)

$d_{\text{perihelion}} = 147.09 \text{ Gm}$  = closest sun-Earth distance, which occurs about 3 January (NASA 2015)

$d_r = 173 = 22 \text{ June}$  = approx. day of summer solstice

$e = 0.0167$  = eccentricity of Earth orbit around sun

$g = -9.80665 \text{ m}\cdot\text{s}^{-2}$  = average gravitational acceleration on Earth at sea level (negative = downward) (from 2014 CODATA)

$|g| = g_0 \cdot [1 + A \cdot \sin^2(\phi) - B \cdot \sin^2(2\phi)] - C \cdot H$

= variation of gravitational-acceleration magnitude with latitude  $\phi$  & altitude  $H$  (in meters) above mean sea level.  $g_0 = 9.7803184 \text{ m}\cdot\text{s}^{-2}$ ,  $A = 0.0053024$ ,  $B = 0.0000059$ ,  $C = 3.086 \times 10^{-6} \text{ s}^{-2}$ .

$M = 5.9726 \times 10^{24} \text{ kg}$  = mass of Earth (NASA 2015)

$P_{\text{earth}} = 365.256 \text{ days}$  = Earth orbital period (2015)

$P_{\text{moon}} = 27.3217 \text{ days}$  = lunar orbital period (2015)

$P_{\text{sidereal}} = 23.9344696 \text{ h}$  = sidereal day = period for one revolution of the Earth about its axis, relative to fixed stars

$R_{\text{earth}} = 6371.0 \text{ km}$  = volumetric average Earth radius (from NASA 2015)

= 6378.1 km = Earth radius at equator

= 6356.8 km = Earth radius at poles

$S = 1367.6 \text{ W}\cdot\text{m}^{-2}$  = solar irradiance (solar constant) at top of atmosphere (NASA 2015)

$\approx 1.125 \text{ K}\cdot\text{m}\cdot\text{s}^{-1}$  = kinematic solar constant (based on mean sea-level density)

$T_e = 254.3 \text{ K}$  = effective radiation emission black-body temperature of Earth system (NASA 2015)

$\Phi_r = 23.44^\circ = 0.4091 \text{ radians}$  = tilt of Earth axis = obliquity relative to the orbital plane (2015)

$\Omega = 0.7292107 \times 10^{-4} \text{ s}^{-1}$  = sidereal rotation frequency of Earth (NASA 2015)

$2\Omega = 1.458421 \times 10^{-4} \text{ s}^{-1}$  = Coriolis factor

$2\Omega / R_{\text{earth}} = 2.289 \times 10^{-11} \text{ m}^{-1}\cdot\text{s}^{-1}$  = beta factor

#### B.4. AIR AND WATER CHARACTERISTICS

- $a = 0.0337 \text{ (mm/day)} \cdot (\text{W/m}^2)^{-1} =$  water-depth evaporation per unit latent-heat flux  
 $B = 3 \times 10^9 \text{ V}\cdot\text{km}^{-1} =$  breakdown potential for dry air  
 $C_{vd} = 717 \text{ J}\cdot\text{kg}^{-1}\cdot\text{K}^{-1} =$  specific heat for dry air at constant volume  
 $C_{pd} = 1003 \text{ J}\cdot\text{kg}^{-1}\cdot\text{K}^{-1} =$  specific heat for dry air at constant pressure at  $-23^\circ\text{C}$   
 $= 1004 \text{ J}\cdot\text{kg}^{-1}\cdot\text{K}^{-1} =$  specific heat for dry air at constant pressure at  $0^\circ\text{C}$   
 $= 1005 \text{ J}\cdot\text{kg}^{-1}\cdot\text{K}^{-1} =$  specific heat for dry air at constant pressure at  $27^\circ\text{C}$   
 $C_{pv} = 1850 \text{ J}\cdot\text{kg}^{-1}\cdot\text{K}^{-1} =$  specific heat for water vapor at constant pressure at  $0^\circ\text{C}$   
 $= 1875 \text{ J}\cdot\text{kg}^{-1}\cdot\text{K}^{-1} =$  specific heat for water vapor at constant pressure at  $15^\circ\text{C}$   
 $C_{liq} = 4217.6 \text{ J}\cdot\text{kg}^{-1}\cdot\text{K}^{-1} =$  specific heat of liquid water at  $0^\circ\text{C}$   
 $C_{ice} = 2106 \text{ J}\cdot\text{kg}^{-1}\cdot\text{K}^{-1} =$  specific heat of ice at  $0^\circ\text{C}$   
 $D = 2.11 \times 10^{-5} \text{ m}^2\cdot\text{s}^{-1} =$  molecular diffusivity of water vapor in air in standard conditions  
 $e_o = 0.611 \text{ kPa} =$  reference vapor pressure at  $0^\circ\text{C}$   
 $k = 0.0253 \text{ W}\cdot\text{m}^{-1}\cdot\text{K}^{-1} =$  molecular conductivity of air at sea level in standard conditions  
 $L_d = 2.834 \times 10^6 \text{ J}\cdot\text{kg}^{-1} =$  latent heat of deposition at  $0^\circ\text{C}$   
 $L_f = 3.34 \times 10^5 \text{ J}\cdot\text{kg}^{-1} =$  latent heat of fusion at  $0^\circ\text{C}$   
 $L_v = 2.501 \times 10^6 \text{ J}\cdot\text{kg}^{-1} =$  latent heat of vaporization at  $0^\circ\text{C}$   
 $n = 3.3 \times 10^{28} \text{ molecules}\cdot\text{m}^{-3}$  for liquid water at  $0^\circ\text{C}$   
 $n_{air} \approx 1.000277 =$  index of refraction for air  
 $n_{water} \approx 1.336 =$  index of refraction for liquid water  
 $n_{ice} \approx 1.312 =$  index of refraction for ice  
 $P_{STP} = 101.325 \text{ kPa} =$  standard sea-level pressure (STP = Standard Temperature & Pressure)  
 $\mathfrak{R}_d = 0.287053 \text{ kPa}\cdot\text{K}^{-1}\cdot\text{m}^3\cdot\text{kg}^{-1} = C_{pd} - C_{vd} = 287.053 \text{ J}\cdot\text{K}^{-1}\cdot\text{kg}^{-1} =$  gas constant for dry air  
 $\mathfrak{R}_v = 461.5 \text{ J}\cdot\text{K}^{-1}\cdot\text{kg}^{-1} =$  water-vapor gas constant  $= 4.61 \times 10^{-4} \text{ kPa}\cdot\text{K}^{-1}\cdot\text{m}^3\cdot\text{g}^{-1}$   
 $Ri_c = 0.25 =$  critical Richardson number (dimensionless)  
 $s_o = 343.15 \text{ m}\cdot\text{s}^{-1} =$  sound speed in standard, calm air  
 $T_{STP} = 15^\circ\text{C} =$  standard sea-level temperature  
 $\varepsilon = 0.622 \text{ g}_{water}\cdot\text{g}_{air}^{-1} = \mathfrak{R}_d / \mathfrak{R}_v =$  gas-constant ratio  
 $\gamma = 0.0004 \text{ (g}_{water}\cdot\text{g}_{air}^{-1})\cdot\text{K}^{-1} = C_p / L_v = 0.4 \text{ (g}_{water}\cdot\text{kg}_{air}^{-1})\cdot\text{K}^{-1} =$  psychrometric constant  
 $\Gamma_d = 9.75 \text{ K}\cdot\text{km}^{-1} = |g|/C_p =$  dry adiabatic lapse rate  
 $\rho_{STP} = 1.225 \text{ kg}\cdot\text{m}^{-3} =$  standard sea-level air density  
 $\rho_{avg} = 0.689 \text{ kg}\cdot\text{m}^{-3} =$  air density averaged over the troposphere (over  $z = 0$  to  $11 \text{ km}$ )

- $\rho_{liq} = 999.84 \text{ kg}\cdot\text{m}^{-3} =$  density of liquid water at  $0^\circ\text{C}$   
 $= 1000.0 \text{ kg}\cdot\text{m}^{-3} =$  density of liquid water at  $4^\circ\text{C}$   
 $= 998.21 \text{ kg}\cdot\text{m}^{-3} =$  density of liquid water at  $20^\circ\text{C}$   
 $= 992.22 \text{ kg}\cdot\text{m}^{-3} =$  density of liquid water at  $40^\circ\text{C}$   
 $= 983.20 \text{ kg}\cdot\text{m}^{-3} =$  density of liquid water at  $60^\circ\text{C}$   
 $= 971.82 \text{ kg}\cdot\text{m}^{-3} =$  density of liquid water at  $80^\circ\text{C}$   
 $= 958.40 \text{ kg}\cdot\text{m}^{-3} =$  density of liquid water at  $100^\circ\text{C}$   
 $\rho_{sea-water} = 1025 \text{ kg}\cdot\text{m}^{-3} =$  avg. density of sea water (sea water contains  $34.482 \text{ g}$  of salt ions per  $\text{kg}$  of water, on average)  
 $\rho_{ice} = 916.8 \text{ kg}\cdot\text{m}^{-3} =$  density of ice at  $0^\circ\text{C}$   
 $\sigma = 0.076 \text{ N}\cdot\text{m}^{-1} =$  surface tension of pure water at  $0^\circ\text{C}$

#### B.5. CONVERSION FACTORS & COMBINED PARAMETERS

- $C_{pd} / C_{vd} = k = 1.400$  (dimensionless) = specific heat ratio  
 $C_{pd} / |g| = 102.52 \text{ m}\cdot\text{K}^{-1}$   
 $C_{pd} / L_v = 0.0004 \text{ (g}_{water}\cdot\text{g}_{air}^{-1})\cdot\text{K}^{-1} = \gamma = 0.4 \text{ (g}_{water}\cdot\text{kg}_{air}^{-1})\cdot\text{K}^{-1} =$  psychrometric constant  
 $C_{pd} / \mathfrak{R}_d = 3.50$  (dimensionless)  
 $C_{vd} / C_{pd} = 1/k = 0.714$  (dimensionless)  
 $|g|/C_{pd} = \Gamma_d = 9.8 \text{ K}\cdot\text{km}^{-1} =$  dry adiabatic lapse rate  
 $|g| / \mathfrak{R}_d = 0.0342 \text{ K}\cdot\text{m}^{-1} = 1/(\text{hypsometric constant})$   
 $L_v / C_{pd} = 2.5 \text{ K}/(\text{g}_{water}\cdot\text{kg}_{air}^{-1})$   
 $L_v / \mathfrak{R}_v = 5423 \text{ K} =$  Clausius-Clapeyron parameter for vaporization  
 $\mathfrak{R}_d / C_{pd} = 0.28571$  (dimensionless) = potential-temperature constant  
 $\mathfrak{R}_d / \mathfrak{R}_v = \varepsilon = 0.622 \text{ g}_{water}\cdot\text{g}_{air}^{-1} =$  gas-constant ratio  
 $\mathfrak{R}_d / |g| = 29.29 \text{ m}\cdot\text{K}^{-1} =$  hypsometric constant  
 $\rho_{air} \cdot C_{pd, air} = 1231 \text{ (W}\cdot\text{m}^{-2}) / (\text{K}\cdot\text{m}\cdot\text{s}^{-1})$  at sea level  
 $= 12.31 \text{ mb}\cdot\text{K}^{-1}$  at sea level  
 $= 1.231 \text{ kPa}\cdot\text{K}^{-1}$  at sea level  
 $\rho_{air} \cdot |g| = 12.0 \text{ kg}\cdot\text{m}^{-2}\cdot\text{s}^{-2}$  at sea level  
 $= 0.12 \text{ mb}\cdot\text{m}^{-1}$  at sea level  
 $= 0.012 \text{ kPa}\cdot\text{m}^{-1}$  at sea level  
 $\rho_{air} \cdot L_v = 3013.5 \text{ (W}\cdot\text{m}^{-2}) / [(\text{g}_{water}\cdot\text{kg}_{air}^{-1})\cdot(\text{m}\cdot\text{s}^{-1})]$  at sea level  
 $\rho_{liq} \cdot C_{liq} = 4.295 \times 10^6 \text{ (W}\cdot\text{m}^{-2}) / (\text{K}\cdot\text{m}\cdot\text{s}^{-1})$   
 $1 \text{ megaton nuclear explosion} \approx 4 \times 10^{15} \text{ J}$   
 $2\pi \text{ radians} = 360^\circ$   
 $(1-\varepsilon)/\varepsilon = 0.61 =$  virtual temperature constant